

Understanding Lead in Tap Water: Chemistry, Control, and Challenges



Stephen J. Randtke, Ph.D., P.E.
Professor
University of Kansas

Rationale

Understanding the sources and forms of lead in drinking water, factors influencing lead release, and the challenges involved can help water utility and public health personnel:

- Diagnose problems
- Identify and evaluate solutions
- Communicate with the public and others
- Avoid unintended consequences
- Avoid future problems

Learning Objectives

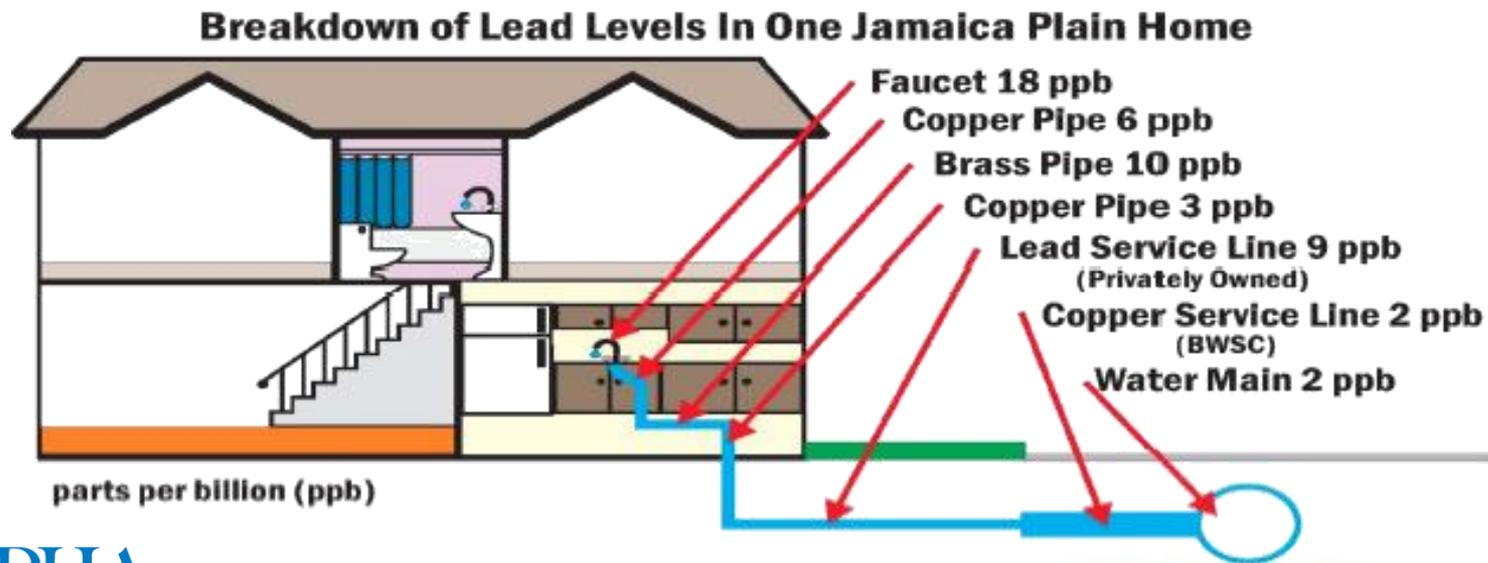
- Identify sources and forms of lead in tap water, and methods for its control.
- Appreciate various challenges involved in controlling lead levels.
- Communicate more effectively with others regarding lead in tap water.

Overview

- Sources & Forms of Lead in Tap Water
- The Role of Water Quality
- Control Options
- Challenges
- Summary & Closing Thoughts

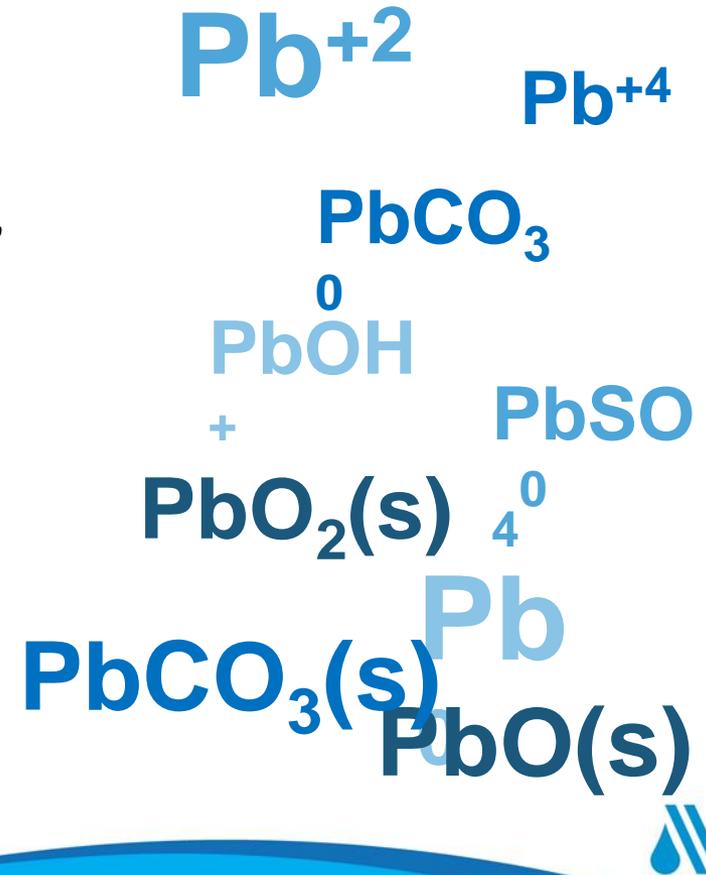
Sources of Lead in Tap Water

- Lead service lines (LSLs)
- Lead solder
- Plumbing components, esp. if brass
- Lead incorporated into scale deposits



Forms of Lead in Tap Water

- Lead may be
 - Dissolved
 - Complexed with carbonate, hydroxide, sulfide, organic material, etc.
 - Composed of, or adsorbed on, corrosion products
 - Lead particles



The Role of Water Quality

Impacts

- Lead solubility
- Lead speciation
- Behavior of pipe scales containing lead



Parameters of interest

- pH, alkalinity, hardness
- Temperature
- Chloride and total dissolved solids (TDS)
- Residual chlorine
- Iron and manganese
- Organic matter
- Stability (chemical and biological)



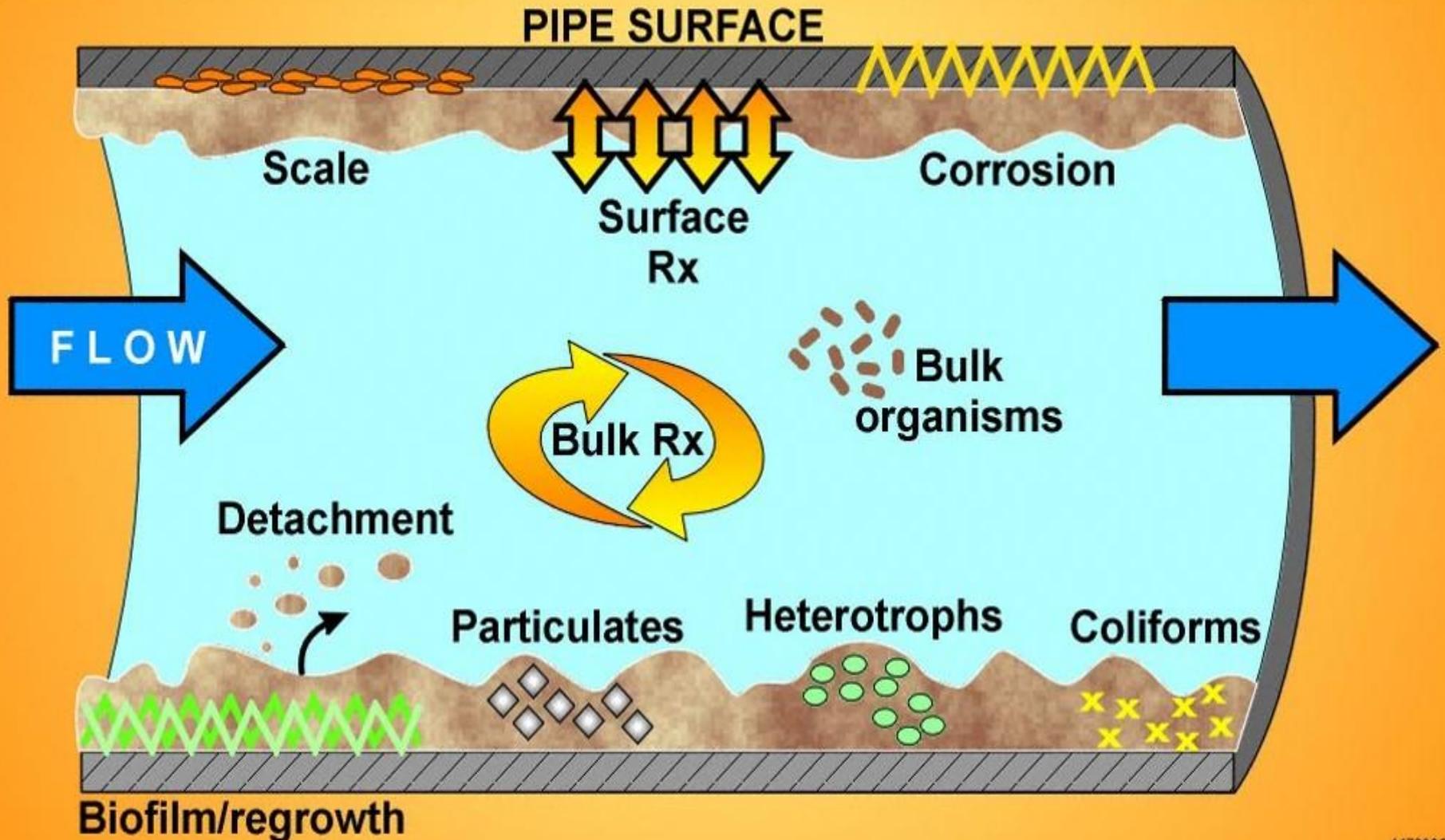
Influences on Water Quality

- Changes in source water quality
- Changes in treatment
- Design and operation of the distribution system:
 - Pipe materials and condition
 - Water age
 - Water disinfection practices
 - Maintenance, e.g., flushing & pigging

Water Main Maintenance



The Distribution System as Reactor



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Control Options

- Corrosion control treatment (CCT)
 - Required for all systems subject to the Lead & Copper Rule (LCR)
 - The two most common methods are:
 - Adjusting pH and alkalinity
 - Orthophosphate addition

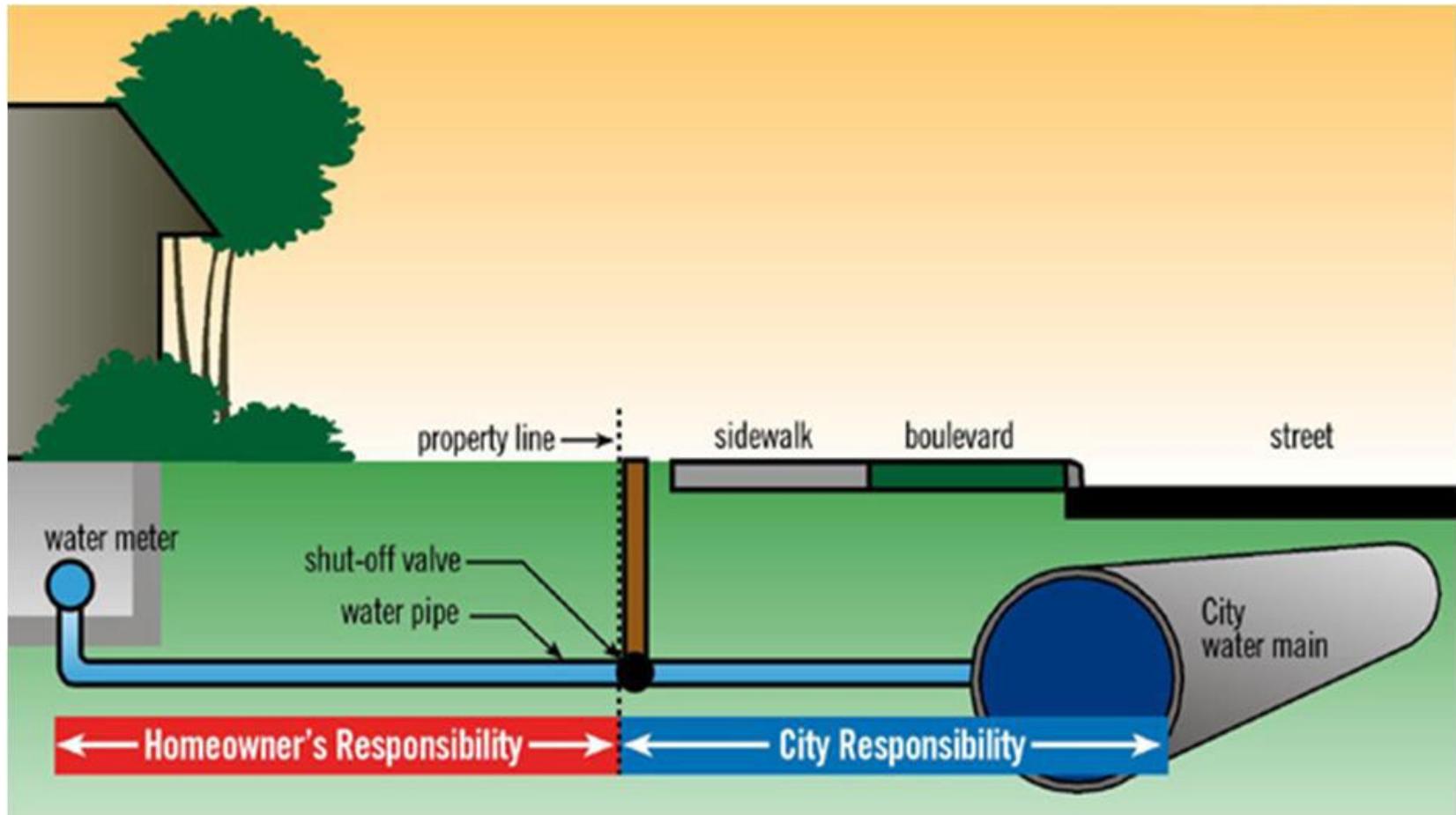
Control Option Selection

- Step 1 - Understand water chemistry
- Step 2 - Evaluate options
- Step 3 - Implement selected option
- Step 4 - Monitor and manage performance

Control Options (cont'd)

- LSL Replacement (LSLR)
 - Partial (PLSLR) or full (FLSLR) replacement
 - Most partial; homeowners reluctant to pay for full
 - Can cause short-term increases in lead levels
 - Expected to be beneficial over time, esp. full
 - Most to date voluntary
 - Proposal to require FLSLR by 2050 (NDWAC, 2015)
 - Noteworthy examples: Madison, Wisc. (mandatory FLSLR); Saskatoon, Sask. (FLSLR mandatory if the City replaces an LSL; voluntary if no problems occur)*

Typical LSL Ownership



Madison's FLSLR Program

- 66,000 connections (est.)
 - Approx. 11,000 LSLs, 5,600 customer-owned
- CCT found to increase lead levels
- City ordinance: MGO Section 13.18
 - All LSLs must be replaced within 10 years; sooner for higher risk sites
 - City to reimburse customer for half their cost, up to \$1,000 (average paid was \$670)
- Completed by Jan. 1, 2011
- Cost ~\$2,985 per FLSLR, incl. reimbursements (\$15.5M total)
- 90th-percentile Pb dropped from ~16 ppb to 2.6–3.6 ppb

Please visit <http://www.cityofmadison.com/water/>, or review the presentation by Grande (ACE 2012) for more information.

Control Options (cont'd)

- Lining or coating LSLs
 - Options include PET linings and epoxy coatings
 - May be advantageous if replacement is difficult
- Options for consumers include:
 - Flush lines (gently) and draw water from the main
 - Install “lead-free” faucets, valves, etc.
 - Install (and maintain) filters certified for removal of the applicable forms of lead (particulate and/or dissolved)

Within the Home

- Is water in the home ... at particular faucets in routine use?
- Are newer lead-free faucets and other fixtures installed?
- Are faucet aerators cleaned regularly?
- Are treatment devices changing water chemistry?

Lead Control Challenges

- Understanding the chemistry
- Monitoring and data collection
- Maintaining water quality
- Regulatory uncertainty
- Public policy — private property tensions
- Communicating effectively with all of the stakeholders
- Balancing competing objectives

Balancing Competing Objectives

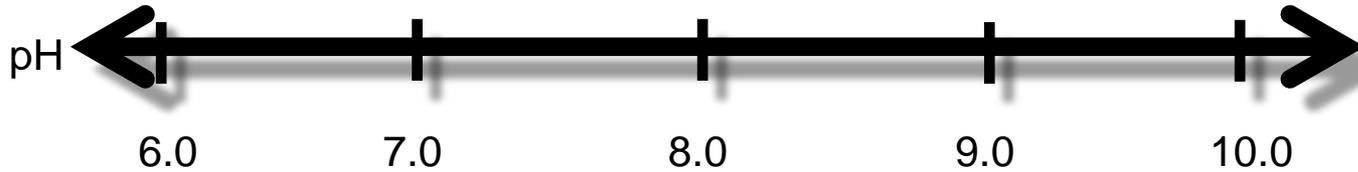
Optimal range for free chlorine disinfection (pH 6.0 – 7.0)

Optimal range for chloramination (pH 8.0 – 9.0)

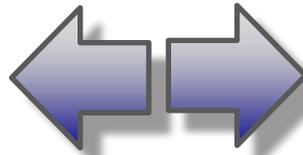
Optimal range for PO₄ (pH 7.2 – 7.8)

Optimal range for Alkalinity/pH Adjustment (pH >9.0)

Historical iron corrosion control



HAA formation increases



THM formation increases

Difficulty reaching CT increases

Lead Control Challenges (cont'd)

- Economic, social, managerial, educational, and other challenges
- Reaching community consensus on a path forward – and deciding who will pay for it!

Summary & Concluding Remarks

- The chemistry of lead in tap water is complex, typically involving multiple sources and forms of lead, with many different factors influencing the levels present in a given sample.
- Controlling lead in tap water can be a challenging task on many different levels.

Summary & Concluding Remarks

- It is important to recognize and appreciate the complex nature of the issue, and the challenges involved, to:
 - Adequately understand the problem
 - Communicate effectively with stakeholders
 - Identify, and reach consensus on, the best option(s) for a given set of circumstances
 - Avoid unintended consequences

Summary & Concluding Remarks

- Do not hesitate to seek help – the sooner the better in most cases!

Acknowledgement

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References & Suggested Reading

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